

1) 8i(7): Most General Relation between ω and Ω .

We have:

$$\Omega = \frac{(\gamma_{nv} + eA) c^2}{\left(r + \frac{eA}{\gamma_{m\omega}}\right) (mc^2(1+\gamma) + e\phi)} \quad - (1)$$

$$\text{So: } \Omega \left(r + \frac{eA}{\gamma_{m\omega}} \right) = x \quad - (2)$$

$$\text{where: } x = \frac{(\gamma_{nv} + eA) c^2}{mc^2(1+\gamma) + e\phi} \quad - (3)$$

$$\text{So: } \boxed{\omega = \frac{eA \Omega}{\gamma_m (x - \Omega r)}} \quad - (4)$$

In the limit:

$$v = 0 \quad - (5)$$

(for an initially stationary electron):

$$x = \frac{eA c^2}{mc^2(1+\gamma) + e\phi} \quad - (6)$$

If we now consider:

$$mc^2(1+\gamma) \gg e\phi \quad - (7)$$

$$\text{and } \gamma \rightarrow 1 \quad - (8)$$

$$x \rightarrow \frac{eA}{2m} \quad - (9)$$

So:

$$\omega \rightarrow \frac{eA\Omega}{m \left(\frac{eA}{2m} - \Omega r \right)} \quad - (10)$$

Finally, if the electron is initially at:

$$r = 0 \quad - (11)$$

Then

$$\omega \rightarrow 2\Omega \quad - (12)$$